

Type's of Dielectrics.

25

Non Polar Dielectric

(Dielectric with non-polar Molecule)

Polar dielectric

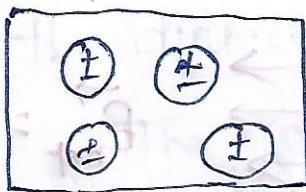
(Dielectric with polar molecule)

→ In non-polar molecule the centre of positive and negative charge coincide. the molecule has no permanent (intrinsic) dipole moment.

Ex:- O_2 , H_2 , Cl_2 etc

Because of their symmetry no dipole moment in absence of external field.

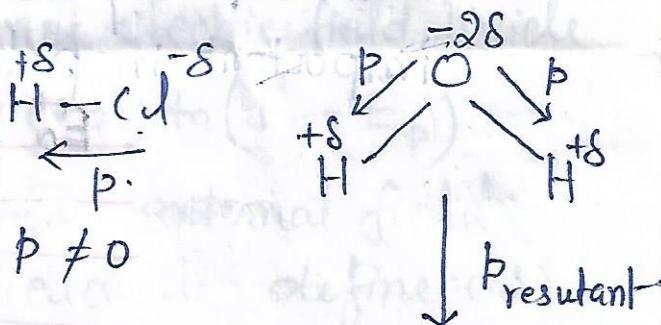
$$\mu = 0$$



→ A polar molecule is one in which the centre of positive and negative charges are separated (even when no external field).

→ These molecules having permanent dipole moment.

Ex:- HCl , H_2O , all polar molecules etc

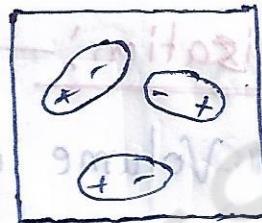


→ In Case of External field (+ve) and (-ve) charges of a non polar molecule are displaced in opposite direction

The displacement stop when external force balanced by restoring force of molecule.

→ In this Case non polar molecule develop induced dipole moment in the direction of field and proportional to field strength.

$$\mu \neq 0$$



→ In polar molecule's also develop a net dipole moment in external field but for different reason.

In absence of external field due to random orientation of dipole moment due to thermal agitation

Polar dielectrics:-

→ When external field applied the individual dipole moment tends to align with the field. The net dipole moment in the direction of external field then it is called.

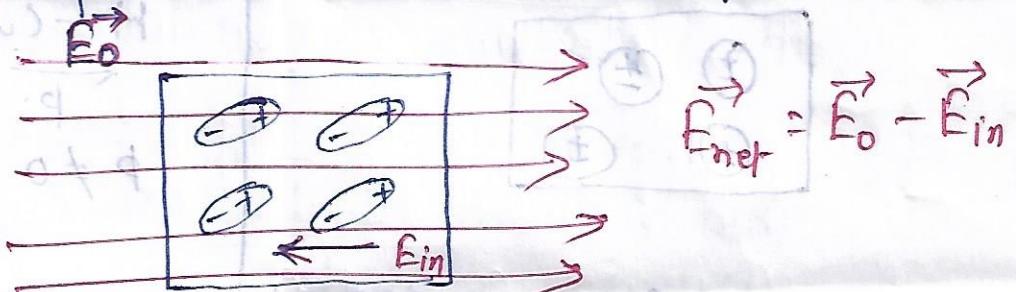
→ ~~The extent of polarization depends~~
Polarised dielectric.

→ The extent of polarization depend on the relative strength of two mutually opposite factors:

(i) The dipole potential energy in the external field tending to align the dipoles with the field.

(ii) Thermal Energy tending to disrupt the alignment.

* Induced dipole moment alignment is more important in polar molecule than non-polar.



* Polarization: → The dipole moment per unit volume of material is called polarization.

It is denoted by "P".

$$P = \frac{p}{\text{Volume}}$$

→ The magnitude of polarization directly proportional to external field.

$$P \propto E$$

$$\boxed{P = \chi_e E}$$

→ Electric Susceptibility.

→ Where proportionality constant is define as Electric Susceptibility.

* ~~that~~.

* Finally we can say that dielectric material oppose to external field i.e. reduces the effect of external field.

* Dielectric Constant : \rightarrow When dielectric material put into external field then net electric field inside the dielectric gets reduced to ($E_0 - E_p$).

The ratio of applied external field and reduced electric field is define as dielectric Constant (K).

$$K = \frac{E_0}{E_0 - E_p}$$

where E_p is polarised Electric Field and E_0 External field.

$$E_0 > (E_0 - E_p)$$

$$\frac{E_0}{E_0 - E_p} > 1 \Rightarrow K > 1$$

Its value always greater than one.

* Dielectric Strength: \rightarrow It is defined as the maximum value of electric field can exist in dielectric without causing the breakdown of its insulating property.

$$E_d = \frac{Q}{A}$$

↳ units in Farad/m

↳ Dielectric strength of materials like glass, mica, air, etc. is about 10^7 to 10^8 N/C.

↳ Dielectric strength of air \approx 3×10^7 N/C.

↳ Dielectric strength of mica \approx 10^8 N/C.

↳ Dielectric strength of glass \approx 10^7 N/C.

↳ Dielectric strength of air \approx 10^7 N/C.

↳ Dielectric strength of mica \approx 10^8 N/C.

↳ Dielectric strength of glass \approx 10^7 N/C.

↳ E_d works

↳ Polarized dielectrics

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$$K = \frac{E_d}{\frac{E_0}{\epsilon - 1}}$$

$$(E_0 \epsilon_0) < E_d$$

$$\boxed{1 < K} \quad \left(\epsilon - 1 < \frac{E_d}{E_0 \epsilon_0} \right)$$